

Lailai Wang, Waihim Kwok and  
Zhongyuan Zhou\*Open Laboratory of Chirotechnology of the  
Institute of Molecular Technology for Drug  
Discovery and Synthesis and Department of  
Applied Biology and Chemical Technology, The  
Hong Kong Polytechnic University, Hung Hom,  
Kowloon, Hong KongCorrespondence e-mail:  
bczyzhou@inet.polyu.edu.hk

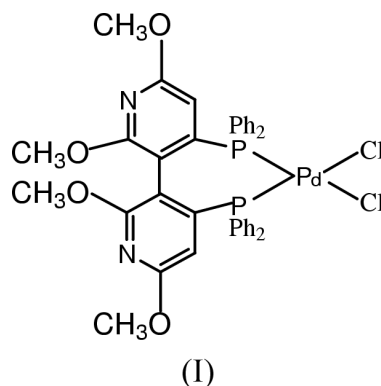
## Key indicators

Single-crystal X-ray study  
 $T = 294$  K  
Mean  $\sigma(\text{C}-\text{C}) = 0.009$  Å  
 $R$  factor = 0.068  
 $wR$  factor = 0.155  
Data-to-parameter ratio = 19.3For details of how these key indicators were  
automatically derived from the article, see  
<http://journals.iucr.org/e>.[(*R*)-4,4'-Bis(diphenylphosphino)-2,2',6,6'-tetra-  
methoxy-3,3'-bipyridine- $\kappa^2P,P'$ ]dichloropalladium(II)

The title compound,  $[\text{PdCl}_2(\text{C}_{38}\text{H}_{34}\text{N}_2\text{O}_4\text{P}_2)]$ , is effective in enantioselective bis-alkoxycarbonylation of styrene. The Pd atoms are located on twofold rotation axes, and there are two independent half-molecules in the asymmetric unit. In the bipyridine systems, the pyridine rings make dihedral angles of  $65.2$  (6) and  $67.9$  (7)° with respect to each other.

## Comment

In recent years, cationic palladium(II) complexes have attracted a great deal of attention in organometallic chemistry, especially because of their catalytic reactions (Drent & Budzelaar, 1996; Sen, 1993) and their application in the self-assembly of various metallamacrocycles (Stang & Olenyuk, 1996). Very recently, we used cationic palladium complexes with novel dipyridylphosphine ligands in the bis-methoxycarbonylation of styrene. In our studies, the enantiomer excess of dimethyl (*R*)-phenylbutanedioates has been improved to 88%. As part of our efforts in investigating these catalytic reactions and the molecular structure of the cation formed by palladium and 4,4'-bis(diphenylphosphine)-2,2',6,6'-tetramethoxy-3,3'-bipyridine (P-Phos), we present the molecular structure of the title compound, (I).

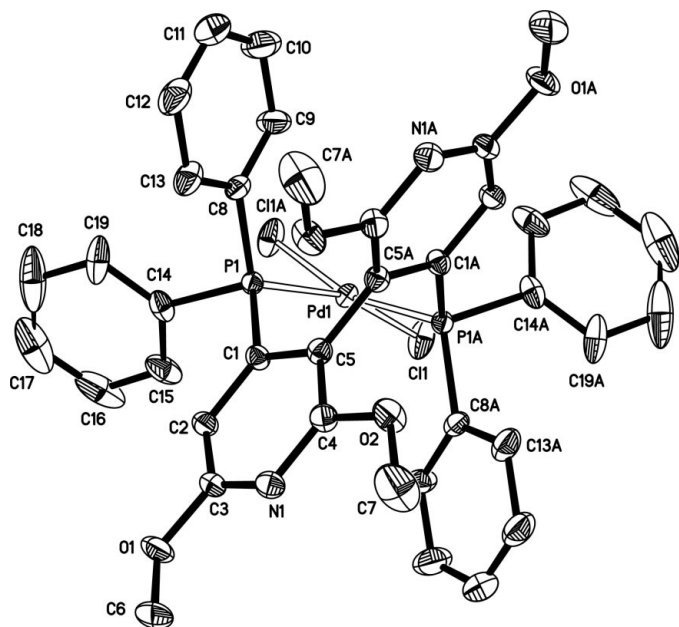


The Pd1 and Pd2 atoms lie on twofold rotation axes. Fig. 1 shows the structure of one of two independent molecules. For the Pd1 complex, the least-squares planes of the two pyridine rings of the bipyridine system in the P-Phos ligand exhibit an interplanar angle of  $65.2$  (6)°, and the C5—C5<sup>i</sup> distance is  $1.511$  (8) Å (see Table 1 for symmetry code). For the Pd2 complex, the corresponding dihedral angle is  $67.9$  (7)° with a C21—C21<sup>ii</sup> distance of  $1.510$  (7) Å.

## Experimental

All reactions and manipulations were carried out under N<sub>2</sub>, using Schlenk techniques or in a glove-box. A solution of [(*R*)-P-Phos]-

Received 13 May 2002  
Accepted 28 May 2002  
Online 8 June 2002



**Figure 1**  
The structure of one of the independent molecules of (I), showing 30% probability displacement ellipsoids.

$\text{Pd}(\text{H}_2\text{O})_2(\text{SO}_3\text{CF}_3)_2$  (87 mg, 0.085 mmol) in methanol (15 ml) was transferred under a nitrogen atmosphere to a 25 ml stainless steel autoclave. The autoclave was heated to 323 K under a CO pressure of 50 bar for 20 h. After the gas was released, the resulting red solution was evaporated *in vacuo*, and the red residue was dried overnight. Red crystals of (I), suitable for X-ray diffraction analysis, were obtained by recrystallization from a solution in  $\text{CHCl}_3$ .  $^{31}\text{P}$  NMR ( $\text{CDCl}_3$ , p.p.m.):  $\delta$  30.6.

#### Crystal data

$[\text{PdCl}_2(\text{C}_{38}\text{H}_{34}\text{N}_2\text{O}_4\text{P}_2)]$   
 $M_r = 821.91$   
 Monoclinic,  $C_2$   
 $a = 19.551(3) \text{ \AA}$   
 $b = 12.1135(19) \text{ \AA}$   
 $c = 18.410(3) \text{ \AA}$   
 $\beta = 119.529(3)^\circ$   
 $V = 3793.7(10) \text{ \AA}^3$   
 $Z = 4$   
 $D_x = 1.439 \text{ Mg m}^{-3}$

Mo  $K\alpha$  radiation  
 Cell parameters from 7972 reflections  
 $\theta = 1\text{--}27.5^\circ$   
 $\mu = 0.76 \text{ mm}^{-1}$   
 $T = 294(2) \text{ K}$   
 Plate, red  
 $0.28 \times 0.24 \times 0.22 \text{ mm}$

#### Data collection

Bruker CCD area-detector diffractometer  
 $\varphi$  and  $\omega$  scans  
 Absorption correction: multi-scan (SADABS; Sheldrick, 1996)  
 $T_{\min} = 0.816$ ,  $T_{\max} = 0.851$   
 12 943 measured reflections

6304 independent reflections  
 5366 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.031$   
 $\theta_{\max} = 27.5^\circ$   
 $h = -25 \rightarrow 24$   
 $k = -15 \rightarrow 9$   
 $l = -23 \rightarrow 23$

#### Refinement

Refinement on  $F^2$   
 $R[F^2 > 2\sigma(F^2)] = 0.068$   
 $wR(F^2) = 0.155$   
 $S = 1.13$   
 8354 reflections  
 433 parameters  
 H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.05P)^2]$   
 where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\max} < 0.001$   
 $\Delta\rho_{\max} = 0.64 \text{ e \AA}^{-3}$   
 $\Delta\rho_{\min} = -0.35 \text{ e \AA}^{-3}$   
 Absolute structure: Flack (1983);  
 1740 Friedel pairs  
 Flack parameter = 0.01 (10)

**Table 1**

Selected geometric parameters ( $\text{\AA}$ ,  $^\circ$ ).

Pd1—P1	2.2504 (11)	Pd2—P2 <sup>ii</sup>	2.2681 (11)
Pd1—Cl1	2.3433 (14)	Pd2—Cl2	2.3167 (12)
C5—C5 <sup>i</sup>	1.511 (8)	C21—C21 <sup>ii</sup>	1.510 (7)
P1 <sup>i</sup> —Pd1—P1	94.57 (6)	P2 <sup>ii</sup> —Pd2—P2	93.26 (6)
P1—Pd1—Cl1 <sup>i</sup>	89.15 (5)	P2—Pd2—Cl2	89.97 (4)
Cl1 <sup>i</sup> —Pd1—Cl1	91.52 (8)	Cl2—Pd2—Cl2 <sup>ii</sup>	87.69 (6)

Symmetry codes: (i)  $-x, y, -z$ ; (ii)  $1-x, y, 1-z$ .

The large elongated ellipsoids of the atom displacement parameters of C16 and C31 suggest orientational disorder of the phenyl groups.

Data collection: *SMART* (Bruker, 1997); cell refinement: *SMART*; data reduction: *SHELXTL-NT* (Bruker, 1997); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL-NT*; software used to prepare material for publication: *SHELXTL-NT*.

#### References

- Bruker (1997). *SMART* (Version 5.0) and *SHELXTL-NT* (Version 5.10). Bruker AXS Inc., Madison, Wisconsin, USA.  
 Drent, E. & Budzelaar, P. H. M. (1996). *Chem. Rev.* **96**, 663–681.  
 Flack, H. D. (1983). *Acta Cryst.* **A39**, 876–881.  
 Sen, A. (1993). *Acc. Chem. Res.* **26**, 303–310.  
 Sheldrick, G. M. (1996). *SADABS*. University of Göttingen, Germany.  
 Sheldrick, G. M. (1997). *SHELXL97* and *SHELXS97*. University of Göttingen, Germany.  
 Stang, P. J. & Olenyuk, B. (1996). *Angew. Chem.* **108**, 798–802.